PERIODIC TABLE

Atomic Properties of the Elements

NATIONAL INSTITUTE OF STANDARDS AND TECHNOLOGY
U.S. DEPARTMENT OF COMMERCE

Physical Measurement Laboratory www.nist.gov/pml
Standard Reference Data www.nist.gov/srd

FREQUENTLY USED FUNDAMENTAL PHYSICAL CONSTANTS§

1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ^{133}Cs

speed of light in vacuum	c	299 792 458 m s^{-1}	(exact)
Planck constant	h	6.626 070 15 x 10^{-34} J Hz^{-1}	(exact)
elementary charge	e	1.602 176 634 x 10^{-19} C	(exact)
Avogadro constant	N_A	6.022 140 76 x 10^{23} mol^{-1}	(exact)
Boltzmann constant	k	1.380 649 x 10^{-23} J K^{-1}	(exact)
electron volt	eV	1.602 176 634 x 10^{-19} J	(exact)
electron mass	m_e	9.109 383 71 x 10^{-31} kg	
energy equivalent	m_ec^2	0.510 998 951 MeV	
proton mass	m_p	1.672 621 926 x 10^{-27} kg	
energy equivalent	m_pc^2	938.272 089 MeV	
fine-structure constant	α	1/137.035 999	
Rydberg energy	$R_\infty hc$	13.605 693 1230 eV	
Newtonian constant of gravitation	G	6.674 x 10^{-11} m^3 kg^{-1} s^{-2}	

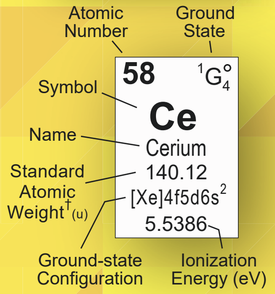
§For the most accurate values of these and other constants, visit pml.nist.gov/constants.

- ☐ Solids
- 🟪 Liquids
- 🟥 Gases
- 🟦 Artificially Prepared

Group

Period	1 IA	2 IIA	3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIII	9	10	11 IB	12 IIB	13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA
1	1 ${}^2S_{1/2}$ **H** Hydrogen 1.008 1s 13.5984																	2 ${}^1S_0$ **He** Helium 4.0026 $1s^2$ 24.5874
2	3 ${}^2S_{1/2}$ **Li** Lithium 6.94 $1s^22s$ 5.3917	4 ${}^1S_0$ **Be** Beryllium 9.0122 $1s^22s^2$ 9.3227											5 ${}^2P^\circ_{1/2}$ **B** Boron 10.81 $1s^22s^22p$ 8.2980	6 ${}^3P_0$ **C** Carbon 12.011 $1s^22s^22p^2$ 11.2603	7 ${}^4S^\circ_{3/2}$ **N** Nitrogen 14.007 $1s^22s^22p^3$ 14.5341	8 ${}^3P_2$ **O** Oxygen 15.999 $1s^22s^22p^4$ 13.6181	9 ${}^2P^\circ_{3/2}$ **F** Fluorine 18.998 $1s^22s^22p^5$ 17.4228	10 ${}^1S_0$ **Ne** Neon 20.180 $1s^22s^22p^6$ 21.5645
3	11 ${}^2S_{1/2}$ **Na** Sodium 22.990 [Ne]3s 5.1391	12 ${}^1S_0$ **Mg** Magnesium 24.305 [Ne]$3s^2$ 7.6462											13 ${}^2P^\circ_{1/2}$ **Al** Aluminum 26.982 [Ne]$3s^23p$ 5.9858	14 ${}^3P_0$ **Si** Silicon 28.085 [Ne]$3s^23p^2$ 8.1517	15 ${}^4S^\circ_{3/2}$ **P** Phosphorus 30.974 [Ne]$3s^23p^3$ 10.4867	16 ${}^3P_2$ **S** Sulfur 32.06 [Ne]$3s^23p^4$ 10.3600	17 ${}^2P^\circ_{3/2}$ **Cl** Chlorine 35.45 [Ne]$3s^23p^5$ 12.9676	18 ${}^1S_0$ **Ar** Argon 39.95 [Ne]$3s^23p^6$ 15.7596
4	19 ${}^2S_{1/2}$ **K** Potassium 39.098 [Ar]4s 4.3407	20 ${}^1S_0$ **Ca** Calcium 40.078 [Ar]$4s^2$ 6.1132	21 ${}^2D_{3/2}$ **Sc** Scandium 44.956 [Ar]$3d4s^2$ 6.5615	22 ${}^3F_2$ **Ti** Titanium 47.867 [Ar]$3d^24s^2$ 6.8281	23 ${}^4F_{3/2}$ **V** Vanadium 50.942 [Ar]$3d^34s^2$ 6.7462	24 ${}^7S_3$ **Cr** Chromium 51.996 [Ar]$3d^54s$ 6.7665	25 ${}^6S_{5/2}$ **Mn** Manganese 54.938 [Ar]$3d^54s^2$ 7.4340	26 ${}^5D_4$ **Fe** Iron 55.845 [Ar]$3d^64s^2$ 7.9025	27 ${}^4F_{9/2}$ **Co** Cobalt 58.933 [Ar]$3d^74s^2$ 7.8810	28 ${}^3F_4$ **Ni** Nickel 58.693 [Ar]$3d^84s^2$ 7.6399	29 ${}^2S_{1/2}$ **Cu** Copper 63.546 [Ar]$3d^{10}4s$ 7.7264	30 ${}^1S_0$ **Zn** Zinc 65.38 [Ar]$3d^{10}4s^2$ 9.3942	31 ${}^2P^\circ_{1/2}$ **Ga** Gallium 69.723 [Ar]$3d^{10}4s^24p$ 5.9993	32 ${}^3P_0$ **Ge** Germanium 72.630 [Ar]$3d^{10}4s^24p^2$ 7.8994	33 ${}^4S^\circ_{3/2}$ **As** Arsenic 74.922 [Ar]$3d^{10}4s^24p^3$ 9.7886	34 ${}^3P_2$ **Se** Selenium 78.971 [Ar]$3d^{10}4s^24p^4$ 9.7524	35 ${}^2P^\circ_{3/2}$ **Br** Bromine 79.904 [Ar]$3d^{10}4s^24p^5$ 11.8138	36 ${}^1S_0$ **Kr** Krypton 83.798 [Ar]$3d^{10}4s^24p^6$ 13.9996
5	37 ${}^2S_{1/2}$ **Rb** Rubidium 85.468 [Kr]5s 4.1771	38 ${}^1S_0$ **Sr** Strontium 87.62 [Kr]$5s^2$ 5.6949	39 ${}^2D_{3/2}$ **Y** Yttrium 88.906 [Kr]$4d5s^2$ 6.2173	40 ${}^3F_2$ **Zr** Zirconium 91.224 [Kr]$4d^25s^2$ 6.6341	41 ${}^6D_{1/2}$ **Nb** Niobium 92.906 [Kr]$4d^45s$ 6.7589	42 ${}^7S_3$ **Mo** Molybdenum 95.95 [Kr]$4d^55s$ 7.0924	43 ${}^6S_{5/2}$ **Tc** Technetium (97) [Kr]$4d^55s^2$ 7.1194	44 ${}^5F_5$ **Ru** Ruthenium 101.07 [Kr]$4d^75s$ 7.3605	45 ${}^4F_{9/2}$ **Rh** Rhodium 102.91 [Kr]$4d^85s$ 7.4589	46 ${}^1S_0$ **Pd** Palladium 106.42 [Kr]$4d^{10}$ 8.3368	47 ${}^2S_{1/2}$ **Ag** Silver 107.87 [Kr]$4d^{10}5s$ 7.5762	48 ${}^1S_0$ **Cd** Cadmium 112.41 [Kr]$4d^{10}5s^2$ 8.9938	49 ${}^2P^\circ_{1/2}$ **In** Indium 114.82 [Kr]$4d^{10}5s^25p$ 5.7864	50 ${}^3P_0$ **Sn** Tin 118.71 [Kr]$4d^{10}5s^25p^2$ 7.3439	51 ${}^4S^\circ_{3/2}$ **Sb** Antimony 121.76 [Kr]$4d^{10}5s^25p^3$ 8.6084	52 ${}^3P_2$ **Te** Tellurium 127.60 [Kr]$4d^{10}5s^25p^4$ 9.0098	53 ${}^2P^\circ_{3/2}$ **I** Iodine 126.90 [Kr]$4d^{10}5s^25p^5$ 10.4512	54 ${}^1S_0$ **Xe** Xenon 131.29 [Kr]$4d^{10}5s^25p^6$ 12.1298
6	55 ${}^2S_{1/2}$ **Cs** Cesium 132.91 [Xe]6s 3.8939	56 ${}^1S_0$ **Ba** Barium 137.33 [Xe]$6s^2$ 5.2117		72 ${}^3F_2$ **Hf** Hafnium 178.49 [Xe]$4f^{14}5d^26s^2$ 6.8251	73 ${}^4F_{3/2}$ **Ta** Tantalum 180.95 [Xe]$4f^{14}5d^36s^2$ 7.5496	74 ${}^5D_0$ **W** Tungsten 183.84 [Xe]$4f^{14}5d^46s^2$ 7.8640	75 ${}^6S_{5/2}$ **Re** Rhenium 186.21 [Xe]$4f^{14}5d^56s^2$ 7.8335	76 ${}^5D_4$ **Os** Osmium 190.23 [Xe]$4f^{14}5d^66s^2$ 8.4382	77 ${}^4F_{9/2}$ **Ir** Iridium 192.22 [Xe]$4f^{14}5d^76s^2$ 8.9670	78 ${}^3D_3$ **Pt** Platinum 195.08 [Xe]$4f^{14}5d^96s$ 8.9588	79 ${}^2S_{1/2}$ **Au** Gold 196.97 [Xe]$4f^{14}5d^{10}6s$ 9.2256	80 ${}^1S_0$ **Hg** Mercury 200.59 [Xe]$4f^{14}5d^{10}6s^2$ 10.4375	81 ${}^2P^\circ_{1/2}$ **Tl** Thallium 204.38 [Hg]6p 6.1083	82 ${}^3P_0$ **Pb** Lead 207.2 [Hg]$6p^2$ 7.4167	83 ${}^4S^\circ_{3/2}$ **Bi** Bismuth 208.98 [Hg]$6p^3$ 7.2855	84 ${}^3P_2$ **Po** Polonium (209) [Hg]$6p^4$ 8.4181	85 ${}^2P^\circ_{3/2}$ **At** Astatine (210) [Hg]$6p^5$ 9.3175	86 ${}^1S_0$ **Rn** Radon (222) [Hg]$6p^6$ 10.7485
7	87 ${}^2S_{1/2}$ **Fr** Francium (223) [Rn]7s 4.0727	88 ${}^1S_0$ **Ra** Radium (226) [Rn]$7s^2$ 5.2784		104 ${}^3F_2$ **Rf** Rutherfordium (267) [Rn]$5f^{14}6d^27s^2$ 6.02	105 ${}^4F_{3/2}$ **Db** Dubnium (268) [Rn]$5f^{14}6d^37s^2$ 6.8	106 $_0$ **Sg** Seaborgium (269) [Rn]$5f^{14}6d^47s^2$ 7.8	107 $_{5/2}$ **Bh** Bohrium (270) [Rn]$5f^{14}6d^57s^2$ 7.7	108 $_4$ **Hs** Hassium (269) [Rn]$5f^{14}6d^67s^2$ 7.6	109 **Mt** Meitnerium (277)	110 **Ds** Darmstadtium (281)	111 **Rg** Roentgenium (282)	112 **Cn** Copernicium (285)	113 **Nh** Nihonium (286)	114 **Fl** Flerovium (290)	115 **Mc** Moscovium (290)	116 **Lv** Livermorium (293)	117 **Ts** Tennessine (294)	118 **Og** Oganesson (294)

Lanthanides	57 ${}^2D_{3/2}$ **La** Lanthanum 138.91 [Xe]$5d6s^2$ 5.5769	58 ${}^1G^\circ_4$ **Ce** Cerium 140.12 [Xe]$4f5d6s^2$ 5.5386	59 ${}^4I^\circ_{9/2}$ **Pr** Praseodymium 140.91 [Xe]$4f^36s^2$ 5.4702	60 ${}^5I_4$ **Nd** Neodymium 144.24 [Xe]$4f^46s^2$ 5.5250	61 ${}^6H^\circ_{5/2}$ **Pm** Promethium (145) [Xe]$4f^56s^2$ 5.5819	62 ${}^7F_0$ **Sm** Samarium 150.36 [Xe]$4f^66s^2$ 5.6437	63 ${}^8S^\circ_{7/2}$ **Eu** Europium 151.96 [Xe]$4f^76s^2$ 5.6704	64 ${}^9D^\circ_2$ **Gd** Gadolinium 157.25 [Xe]$4f^75d6s^2$ 6.1498	65 ${}^6H^\circ_{15/2}$ **Tb** Terbium 158.93 [Xe]$4f^96s^2$ 5.8638	66 ${}^5I_8$ **Dy** Dysprosium 162.50 [Xe]$4f^{10}6s^2$ 5.9391	67 ${}^4I^\circ_{15/2}$ **Ho** Holmium 164.93 [Xe]$4f^{11}6s^2$ 6.0215	68 ${}^3H_6$ **Er** Erbium 167.26 [Xe]$4f^{12}6s^2$ 6.1077	69 ${}^2F^\circ_{7/2}$ **Tm** Thulium 168.93 [Xe]$4f^{13}6s^2$ 6.1844	70 ${}^1S_0$ **Yb** Ytterbium 173.05 [Xe]$4f^{14}6s^2$ 6.2542	71 ${}^2D_{3/2}$ **Lu** Lutetium 174.97 [Xe]$4f^{14}5d6s^2$ 5.4259
Actinides	89 ${}^2D_{3/2}$ **Ac** Actinium (227) [Rn]$6d7s^2$ 5.3802	90 ${}^3F_2$ **Th** Thorium 232.04 [Rn]$6d^27s^2$ 6.3067	91 ${}^4K_{11/2}$ **Pa** Protactinium 231.04 [Rn]$5f^26d7s^2$ 5.89	92 ${}^5L^\circ_6$ **U** Uranium 238.03 [Rn]$5f^36d7s^2$ 6.1941	93 ${}^6L_{11/2}$ **Np** Neptunium (237) [Rn]$5f^46d7s^2$ 6.2655	94 ${}^7F_0$ **Pu** Plutonium (244) [Rn]$5f^67s^2$ 6.0258	95 ${}^8S^\circ_{7/2}$ **Am** Americium (243) [Rn]$5f^77s^2$ 5.9738	96 ${}^9D^\circ_2$ **Cm** Curium (247) [Rn]$5f^76d7s^2$ 5.9922	97 ${}^6H^\circ_{15/2}$ **Bk** Berkelium (247) [Rn]$5f^97s^2$ 6.1979	98 ${}^5I_8$ **Cf** Californium (251) [Rn]$5f^{10}7s^2$ 6.2819	99 ${}^4I^\circ_{15/2}$ **Es** Einsteinium (252) [Rn]$5f^{11}7s^2$ 6.3684	100 ${}^3H_6$ **Fm** Fermium (257) [Rn]$5f^{12}7s^2$ 6.50	101 ${}^2F^\circ_{7/2}$ **Md** Mendelevium (258) [Rn]$5f^{13}7s^2$ 6.58	102 ${}^1S_0$ **No** Nobelium (259) [Rn]$5f^{14}7s^2$ 6.6262	103 ${}^2P^\circ_{1/2}$ **Lr** Lawrencium (262) [Rn]$5f^{14}7s^27p$ 4.96

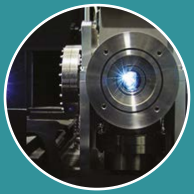


†Based upon ^{12}C. () indicates the mass number of the longest-lived isotope.

For the most precise values and uncertainties visit ciaaw.org and pml.nist.gov/data.

NIST SP 966 (June 2024)

NISTory of the Periodic Table



Deuterium

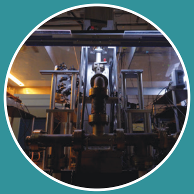
This rare heavy isotope of hydrogen was distilled from liquid hydrogen at NIST and identified by Columbia University's Harold Urey (Nobel Prize 1934).

Image Credit: Uwe Arp/NIST

Krypton

The wavelength of light from this atom was used to define the official meter until 1983.

Image Credit: Neil Tucker/Wikimedia



Cesium

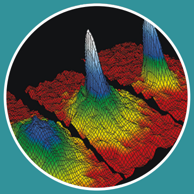
The frequency of light absorbed by this atom, measured by atomic clocks such as NIST-F4, has been used to officially define the second since 1967.

Image Credit: NIST

Sodium

A gas of these atoms was cooled with lasers by NIST scientists to reach temperatures near absolute zero (Nobel Prize 1997).

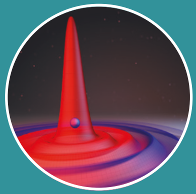
Image Credit: H. Mark Helfer/NIST



Rubidium

Researchers at JILA (NIST-CU Boulder) used these atoms to create a new state of matter called a Bose-Einstein condensate (Nobel Prize 2001).

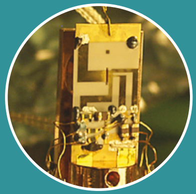
Image Credit: NIST/JILA/CU-Boulder



Potassium

JILA researcher Debbie Jin and her colleagues coaxed pairs of these atoms into forming another new state of matter known as a fermionic condensate.

Image Credit: JILA



Aluminum

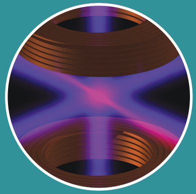
Electrically charged versions of these atoms (ions) have been used to create "quantum logic" clocks with record accuracy.

Image Credit: NIST

Beryllium

Ions of these atoms have performed quantum logic operations that could lay groundwork for future quantum computers.

Image Credit: S. Burrows/JILA



Strontium and Ytterbium

NIST and JILA researchers trapped thousands of these atoms in webs of light known as optical lattices to create ultraprecise and stable atomic clocks.

Image Credit: NIST

Charlotte Moore Sitterly

From 1945 to 1985, this NIST astrophysicist published critically reviewed tables of atomic data, establishing the agency as an authoritative source of this information.

Image Credit: NIST

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